

Energy and Exergy Cost Analysis of Two Different Routes for Vinasse Treatment with Energy Recovery

Milagros Cecilia Palacios-Bereche, Reynaldo Palacios-Bereche and Silvia Azucena Nebra

EasyChair preprints are intended for rapid dissemination of research results and are integrated with the rest of EasyChair.

August 23, 2020

# Energy and exergy cost analysis of two different routes for vinasse treatment with energy recovery

### Milagros Cecilia Palacios-Bereche<sup>a</sup>, Reynaldo Palacios-Bereche<sup>b</sup> and Silvia Azucena Nebra<sup>c</sup>

 <sup>a</sup> Energy Engineering Modeling and Simulation Laboratory, Federal University of ABC, Santo André, S.P., Brazil, milagros.palacios@ufabc.edu.br
 <sup>b</sup> Federal University of ABC, Santo André, S.P., Brazil, reynaldo.palacios@ufabc.edu.br
 <sup>c</sup> Federal University of ABC, Santo André, S.P., Brazil/University of Campinas, Campinas, S.P., Brazil, silvia.nebra@ufabc.edu.br, CA

#### Abstract:

The vinasse, produced as the bottom product of the distillation column of the ethanol production process, is the main liquid residue of this industry. It is a dark brown liquid of acidic nature and high organic matter content, thus making it a polluting effluent. Currently, it is used to fertilise and irrigate sugarcane fields, taking advantage of its nutrients and high water content. However, its disposition is still a problem because of its high production rate, which ranges from 10 to 15 litres of vinasse per litre of ethanol produced. This way, this work addresses the vinasse problem by a preliminary exergy cost analysis of three alternatives for vinasse disposition with energy recovery and a Base Case for comparison purposes; being the analysed cases: i) a Base Case (conventional production process), ii) the vinasse concentration with subsequent incineration, iii) the vinasse biodigestion with burning of the produced biogas in the boiler of the cogeneration system, and iv) the vinasse biodigestion and subsequent biogas purification aiming at the biomethane production. The preliminary results show the clean biogas of Case iii as the product with the highest unit exergy cost (7.03), followed by the biomethane of Case iv (with a unit exergy cost of 6.95), indicating that important irreversibilities are associated to the biogas production route.

#### Keywords:

Vinasse, Concentration, Incineration, Biodigestion, Exergy cost.

## 1. Introduction

Brazil is the biggest producer of sugarcane in the world [1]; and the sugar and ethanol industry is one of the most important sectors of the national economy. The Brazilian ethanol owes its success to its economic competitiveness, which was achieved through economies of scale and technological advances over time [2]. Nowadays, the Brazilian government is presenting the RenovaBio Program, seeking to expand the biofuel production [3], being the ethanol among the biofuels contemplated. Nonetheless, the main liquid residue of this production process, or vinasse, still represents a problem for the industry, because of its difficult and costly disposition due to the large generated volume. Furthermore, with an increasing ethanol production, encouraged by the RenovaBio Program, the vinasse generated will increase as well.

The vinasse, produced as the bottom product of the distillation column, is a dark brown liquid of acidic nature and high organic matter content, thus making it a polluting effluent. However, its solid content is also rich in nutrients such as potassium, sodium, calcium, phosphorous, manganese and nitrogen, among others, which can be used as fertilisers. This way, the fertirrigation, current vinasse disposition, takes advantage of the nutrients in the solid content, and the high water content to fertilise and irrigate at the same time by aspersing the vinasse over the sugarcane crops [4]. Still, the main problem for its disposition is the high production rate, which ranges from 10 to 15 litres of vinasse per litre of ethanol produced [4].

This way, this work addresses the vinasse problem by a preliminary exergy cost analysis of three alternatives for vinasse disposition with energy recovery. Being these alternatives: a) vinasse concentration with subsequent incineration, b) vinasse biodigestion with burning of the produced biogas in the boiler of the cogeneration system, and c) vinasse biodigestion with subsequent biogas purification aiming at the production of biomethane.

This type of analysis (exergoeconomics) is a tool to identify the location, magnitude and source of thermodynamic losses (irreversibilities) in an energy system. Furthermore, it calculates the cost associated with the exergy destruction and exergy losses; besides assessing the production costs of each product in an energy-conversion system that has more than one product. The exergoeconomics is also used to compare technical alternatives and facilitates feasibility and optimization studies [5].

## 2. Processes description and cases

# 2.1. Case i (Base Case): Conventional ethanol, sugar and electricity production process

A conventional ethanol, sugar and electricity production process was considered as a Base Case, for comparison purposes. The conventional process comprises the following sub-systems: sugarcane cleaning and juice extraction, juice treatment for sugar and ethanol production routes, juice concentration, sugar crystallisation, sugar drying, must preparation and cooling, fermentation, and distillation and rectification. The Base Case was assumed as a medium size plant processing 500 tonnes of sugarcane per hour, and consuming saturated steam at 2.5 bar for thermal requirements. Figure 1 presents a simplified flowsheet of the Base Case.



Fig. 1. Flowsheet of Case i: Base Case. Modified from [6].

### 2.2. Case ii: Vinasse concentration with subsequent incineration

The vinasse concentration and incineration route, or Case ii, was considered as a Base Case coupled to a vinasse concentration system, sending the concentrated vinasse to the boiler of the cogeneration system. A seven-effect evaporation system and a concentration up to 65 Brix was considered, as some manufactures already commercialise this type of vinasse evaporators [7]. Figure 2 shows the simplified flowsheet of Case ii, while Figure 3 presents a simplified scheme of the vinasse concentration system.



Fig. 2. Flowsheet of Case ii: Base Case + vinasse concentration system. Modified from [6].



Fig. 3. Scheme of the seven-effect evaporation system for vinasse.

### 2.3. Case iii: Biodigestion of vinasse

This case assumes the vinasse biodigestion to produce biogas, which is send to be burnt in the boiler of the cogeneration system; as can be observed in Figure 4. Mass and energy balances were carried out utilising the software  $\text{EES}^{\textcircled{R}}$  according to [8]. The biogas cleaning was assumed to be carried out in a desulphurisation system according to the THIOPAQ process. The parameters and guidelines for biogas production and desulphurisation were taken from [9]. Figure 4 shows the flowsheet for Case iii.



*Fig. 4. Flowsheet of Case iii: Base Case + biodigestion + desulphurisation systems. Modified from* [6].

### 2.4. Case iv: Biomethane production from vinasse biodigestion

Finally, Case iv assumed a Base Case coupled to a biodigestion system, a desulphurisation system for biogas cleaning, and a purification system for biomethane production. In the same way as in Case iii, the THIOPAQ process was assumed for the desulphurisation process, while the water scrubbing process was selected for biogas purification The parameters and guidelines for biogas production, desulphurisation, and biogas purification were taken from [9]. Figure 5 depicts the flowsheet of Case iv.



*Fig. 5. Flowsheet of Case iii: Base Case + biodigestion + dessulphurisation + purification systems. Modified from* [6].

Figure 6 shows the flowsheet for the biogas purification system with water scrubbing according to [10].



Fig. 6. Flowsheet of purification system with water scrubbing. Source [9].

### 2.5. Cogeneration system

In sugarcane-processing plants, a power steam cycle, using sugarcane bagasse as fuel, is commonly used as the cogeneration system. This cogeneration system was based on a Rankine cycle, producing steam at 65 bar and 520°C [11]. It supplies the requirements of steam, electricity and/or mechanical work to the plant. A configuration assuming condensing-extraction steam turbines (CEST), burning all the available bagasse to maximise the electricity surplus, was adopted. Figure 7 presents the scheme of the configuration system adopted.



Fig. 7. Configuration of the cogeneration system (CEST).

# 3. Methodology

The main steps performed in the present work are listed below:

- Modelling and simulation of conventional production process (Base Case), alternative technologies for vinasse energy use (Cases ii, iii and iv), and cogeneration system;
- exergy analysis;
- exergy cost assessment.

# 3.1. Conventional process, alternative technologies, and cogeneration system simulation

The software Aspen  $\text{Plus}^{\text{TM}}$  V9 was used to simulate the conventional process of sugar, ethanol and electricity production, thus performing mass and energy balances. The simulation was performed according to previous studies [8,12]. Data from the literature were collected to perform the process simulation, and the guidelines from [13–15] were followed. A flowsheet diagram of the simulated conventional process is presented in Figure 8.



*Fig. 8. Sugar and ethanol conventional production process simulated in Aspen Plus*<sup>™</sup> *V9* 

The alternative technologies considered in this study, comprising vinasse concentration, vinasse biodigestion, biogas cleaning and biomethane production, were simulated through mass and energy balances using the software EES<sup>®</sup>; because of its faster convergence and ease of use.

The vinasse concentration and biodigestion were modelled according to previous works [8,12], while the biogas cleaning and biomethane production followed the guidelines presented in [9].

A Rankine cycle was used to model the cogeneration system, considering a steam production at 65 bar and 520°C [11].

The main parameters for the process simulation are presented in Table 1.

Table 1. Main parameters for process simulation

Parameter	Value
Conventional process simulation	
Sugarcane processed, t cane/h	500 <sup>(1)</sup>
Produced sugar, t/h	34.2 <sup>(2)</sup>
Produced hydrous ethanol, m <sup>3</sup> /h	21.1 (2)
Total produced bagasse (50% of humidity); t/h	136 (2)
Steam consumption (sat. @ 2.5 bar) in conventional process, kg/t cane	429.5 <sup>(2)</sup>
Electricity consumption in conventional process, kWh/t cane	28 <sup>(1)</sup>
Vinasse concentration	
Initial brix of vinasse, %	4.29 <sup>(2)</sup>
Final brix of vinasse, %	65 <sup>(3)</sup>
Effect pressures of vinasse concentration system (3)	
Effect 1, bar	2.139
Effect 2, bar	1.788
Effect 3, bar	1.449
Effect 4, bar	1.12
Effect 5, bar	0.802
Effect 6, bar	0.496
Effect 7, bar	0.2
Vinasse biodigestion	
Vinasse COD, kg/m <sup>3</sup>	23.8 <sup>(4)</sup>
COD removal efficiency, %	80 (5)
Biogas conversion factor, Nm <sup>3</sup> <sub>biogas</sub> /kg COD	0.5 (5)
Biogas density, kg/Nm <sup>3</sup>	0.784 (6)
Electricity consumption in biodigestion, kWh/day	230 (6)
Biogas composition <sup>(7)</sup>	
CH <sub>4</sub> , %mol (dry basis)	60
CO <sub>2</sub> , %mol (dry basis)	38.1
H <sub>2</sub> S, %mol (dry basis)	1.9
$H_2O$ , % mol	5.5
Biogas cleaning	
CH <sub>4</sub> concentration increase, %	10.33 (8)
H <sub>2</sub> S removal efficiency, %	99.89 <sup>(8)</sup>
Electricity consumption, kWh/Nm <sup>3</sup> <sub>biogas</sub>	0.024 (8)
Biomethane production	
Methane recovery efficiency, %	99.77 <sup>(8)</sup>
CO <sub>2</sub> separation efficiency, %	98 <sup>(8)</sup>
Vair/V <sub>clean biogas</sub>	2.026 (8)
Water loss, %	2.4 (9)
Electricity consumption, kWh/Nm <sup>3</sup> <sub>clean biogas</sub>	0.2374 (8)
Cogeneration	
Boiler pressure, bar	65 <sup>(10)</sup>
Boiler temperature, °C	520 (10)
Process steam pressure, bar	2.5 (2)
Bagasse LHV (with 50% of humidity), MJ/kg	7.645 (11)
Vinasse HHV (dry), MJ/kg	13.2 (12)
Boiler efficiency, %	85 <sup>(13)</sup>
Turbine efficiency, %	80 (14)
Pump efficiency, %	80 (14)

<sup>(1)</sup> Pina et al. [6]; <sup>(2)</sup> from simulation; <sup>(3)</sup> Fukushima [16]; <sup>(4)</sup> Elia Neto and Shintaku [17]; <sup>(5)</sup> Elia Neto and Shintaku [18]; <sup>(6)</sup> Salomon et al. [19]; <sup>(7)</sup> Leme and Seabra [20]; <sup>(8)</sup> Flores-Zavala [9]; <sup>(9)</sup> Cozma et al. [21]; <sup>(10)</sup> Sosa-Arnao [11]; <sup>(11)</sup> calculated; <sup>(12)</sup> Gallego-Ríos [22]; <sup>(13)</sup> Cortes-Rodríguez et al. [23]; <sup>(14)</sup> Ensinas [15]

### 3.2. Exergy calculation

The exergy of each stream of the evaluated processes was calculated according to previous studies [8,12]. A reference level was chosen at 25°C and 1.01325 bar, according to [24]. The total thermal exergy ( $ex_{tot}$ ) was calculated as the sum of the physical ( $ex_{phy}$ ) and chemical ( $ex_{ch}$ ) exergies [24]:

$$ex_{tot} = ex_{phy} + ex_{ch}.$$
 (1)

The physical exergy was calculated according to (2), neglecting the potential and kinetic components:

$$ex_{phy} = h - h_0 - T_0(s - s_0), \quad (2)$$

where the subscript 0 indicated the reference level.

The chemical exergy is calculated, generally, considering the activity of the stream, as can be observed in (3), considering the standard chemical exergy of pure components (first term) and the losses of chemical exergy due to the dissolution process (second term), according to [24]:

$$ex_{ch} = \left(\frac{1}{\overline{M}}\right) \cdot \left[\sum_{i=1}^{n} y_i \cdot ex_i^{\circ} + \overline{R}_u \cdot T_0 \sum_{i=1}^{n} y_i \cdot \ln(a_i)\right].$$
 (3)

Nevertheless, other approaches were followed for certain streams. Thus, when sucrose-containing streams were contemplated (sugarcane, bagasse, juice, syrup, molasses, sugar), the specific exergy was calculated according to the guidelines presented in [25]. On the other hand, for ethanol-containing streams, the guidelines in [26] were followed. The vinasse, as it is produced and concentrated, was considered as a technical fuel that contains small amounts of sulphur and ashes [24] to calculate its chemical exergy, as in previous studies [8,12]. The streams participating in the biodigestion route (biogas, clean biogas and biomethane) were considered as ideal solutions.

### 3.2. Exergy cost assessment

Since the exergy is an objective measure of the thermodynamic value of an energy carrier, it is also closely related to the economic value of said carrier, because users pay for the potential of energy to cause changes [5]. Thus, the exergoeconomic approach was utilised, since it integrates thermodynamic and economic analysis through the exergy costing, which is the assignment of costs to the exergy content of an energy carrier [5]. The Theory of Exergetic Cost [27] was followed to perform the exergy cost assessment in this study.

An exergetic cost balance was performed in each sub-system of the production process of the proposed cases (4), to calculate the exergetic cost of a flow:

$$\sum \dot{B}_{in} = \sum \dot{B}_{out}, \qquad (4)$$

where  $\dot{B}$  represents the exergetic cost of each flow that enters (*in*) to, and goes out (*out*) from the control volume.

According to [27], the exergetic cost of a flow  $(\dot{B})$  is defined as the amount of exergy required to produce said flow (5):

 $\dot{B}_i = k_i \cdot \dot{E}x_i$ , (5)

where the exergetic cost of an *i* stream is determined by its unit exergetic cost  $(k_i)$  and its total exergy  $(\vec{E}x_i)$ . The total exergy of a stream is calculated by its specific exergy (calculated in the previous section) and the mass flow of the stream, which is given by the process simulation.

Applying (4) to all the sub-systems of the production processes of all the considered cases results in a system of linear equations, where the unit exergetic cost  $(k_i)$  remains unknown. Thus, assumptions were made by following the propositions of the Theory of the Exergetic Cost [27], resulting in additional equations that are required to resolve the equation system.

• A unitary value is assigned as the unit exergy cost  $(k_i)$  of external inputs (sugarcane, freshwater, chemicals).

 $k_{external input} = 1$ .(6)

• By-products of the control volume are assigned a unit exergy cost  $(k_i)$  equal to the input (P4a).

$$k_{by-product} = k_{input} \tag{7}$$

• If a control volume has two or more product streams, then the same unit exergy  $cost(k_i)$  is assigned to all of them (P4b).

$$k_{product1} = k_{product2} = \dots = k_{productn}$$
(8)

• The unit exergy cost  $(k_i)$  of the energy carrier (steam, condensates, vapour bleeds) is determined during its generation (at the boiler of the cogeneration system) and do not change throughout the process.

$$k_{live steam} = k_{processsteam} = k_{condensate} = k_{vapourbleeds}$$
.(9)

• The cost of the irreversibility associated with the operation of the condenser in the cogeneration system, is added to the turbine control volume, thus increasing the unit exergy cost  $(k_i)$  of the electricity.

## 4. Results and discussion

The results of the main products and by-products obtained from the simulation of the evaluated cases are presented in Table 2.

Product/By-product	Case i:	Case ii: Conc. +	Case iii: Biogas+	Case iv: Biomathana
	Base Case	Incineration	burning	Diomeinane
Sugarcane rate, t/h	500	500	500	500
Sugar, kg/t cane	68.4	68.4	68.4	68.4
Hydrous ethanol, l/t cane	42.1*	42.1*	42.1*	42.1*
Bagasse produced in mills, kg/t cane (50% of moisture content)	272	272	272	272
Vinasse, l/t cane	495.6**	495.6**	495.6**	495.6**
Vinasse/ethanol ratio	11.7	11.7	11.7	11.7
Concentrated vinasse, kg/t cane	-	29.9***	-	-
Biogas production, Nm <sup>3</sup> /t cane	-	-	4.72	4.72
Biogas mass flow, kg/t cane	-	-	3.6	3.6
Biodigested vinasse, kg/t cane	-	-	449	449
Clean biogas, kg/t cane	-	-	3.14	3.14
Biomethane, kg/t cane	-	-	-	1.26
Electricity surplus, kWh/t cane	81.2	89.7	85.7	80.4

### Table 2. Main results from simulations

\* at 35°C; \*\* at 74.9°C; \*\*\* at 65 brix

Table 3 presents the results of the energy analysis of the cogeneration system. It can be observed that Case ii and Case iii present higher electricity production than the other cases, being Case ii the one with the highest value. This can be explained because of the additional fuels (concentrated vinasse and clean biogas) that are burned in the boiler, which increase the generated steam (the generated steam of Case ii being 12% higher than the Base Case and Case iv, while an increase of 2.9% for Case iii was obtained), thus increasing the amount of steam that is expanded in the turbine. Moreover, the energy contained in concentrated vinasse was higher, allowing a higher steam production (being the generated steam of Case ii 8.8% higher than Case iii).

On the other hand, since in Case iv the biomethane is considered as an added value product suitable for sale, the only fuel used in the cogeneration system is the bagasse. For this reason, the amount of generated steam is the same as in the Base Case, and the electricity surplus resulted in a lower value because of the additional electricity consumption for this process. Nevertheless, this decrease in electricity surplus is not significant, being only 0.98% lower than the Base Case.

The electricity consumption in Cases iii and iv was higher in comparison to the others cases (i and ii) because of the consumption of the biodigestion, desulphurisation and purification processes, however, this increase was not significant (0.02 kWh/t of cane in Case iii and 0.71 kWh/t of cane in Case iv).

Parameter	Case i: Base Case	Case ii: Conc. + Incineration	Case iii: Biogas + burning	Case iv: Biomethane
Steam: Generation and consumption				
Generated steam in boiler <sup>1</sup> , kg/t cane	552.2	618.6	568.3	552.2
Increasing of generated steam <sup>1</sup> due to new technology, kg/t cane	-	66.4	16.1	-
Steam consumption <sup>2</sup> for vinasse concentration, kg/tcane	-	96.8	-	-
Total steam consumption <sup>2</sup> , kg/t cane	429.5	526.2	429.5	429.5
Fuel used in cogeneration system				
Bagasse, kg/t cane	253.4	253.4	253.4	253.4
Vinasse <sup>3</sup> , kg/t cane	-	29.9	-	-
Clean biogas kg/t cane	-	-	3.14	-
Electricity				
Electricity consumption; kWh/t cane	28	28	28.02	28.71
Electricity surplus, kWh/t cane	81.2	89.7	85.7	80.4

*Table 3. Main results – Cogeneration: CEST Configuration* 

<sup>1</sup> at 520°C and 65 bar; <sup>2</sup> saturated at 2.5 bar; <sup>\*\*\*</sup> at 65 brix

Table 4 and Figure 9 show the main results of the exergy cost assessment, presenting the unit exergy cost of the main products of the evaluated cases. The unit exergy costs for electricity resulted in the range of 4.2 and 4.9; while the unit exergy costs for steam resulted between 3.4 and 3.9.

Regarding the main products, Case ii presented slightly higher unit exergy costs in comparison to the Base Case, because of the higher irreversibilities present in the first one. Furthermore, the unit exergy costs of the products in Case iii were even higher than in Case ii, because the clean biogas used in boiler has a significant unit exergy cost (7.03).

The clean biogas unit exergy cost in Case iii resulted in a higher value than the respective cost of the same product in Case iv, because of the high electricity cost in Case iii.

Regarding the results of Case iv, the unit exergy costs of conventional products resulted the same as in the Base case. The most expensive product in this case is the biomethane, with an exergy cost of 6.9, followed by the electricity and steam, these results show the influence of irreversibilities caused by the biochemical reactions inherent to the biogas production.

Regarding the unit exergy cost of the vinasse that leaves the distillation column, Cases i and iv presents the same value, since both cases presented the same electricity and steam costs, as previously explained. On the other hand, it can be observed that the vinasse unit exergy cost in Cases ii and iii present a higher value; because larger costs of inputs process: electricity and steam, due to additional fuels (concentrated vinasse and biogas), with larger unit exergy costs, are used in boilers.

It is worth mentioning that the cost distribution of The Theory of Exergetic Cost [27], penalise the products at the end of the productive process, accumulating exergy cost [28]. Such is the case of the

biodigestion route, whose products (biogas, clean biogas and biomethane) carry not only the irreversibilities of their respective unit (biodigestion, desulphurisation and purification), but also the irreversibilities of the rest of the process in the vinasse.

Table 4.	Main results – Exergy cost assessment
----------	---------------------------------------

Product	Case i: Case ii: Conc Base Case Incin.		Case iii: Biogas + burning	Case iv: Biomethane	
Ethanol	2.06	2.10	2.14	2.06	
Sugar	1.55	1.57	1.61	1.55	
Vinasse (as produced)	1.80	1.82	1.84	1.80	
Concentrated vinasse (65 brix)	-	1.97	-	-	
Biogas	-	-	6.37	6.21	
Clean biogas	-	-	7.03	6.85	
Biomethane	-	-	-	6.95	
Steam	3.42	3.68	3.90	3.42	
Electricity	4.29	4.58	4.90	4.29	



Fig. 9. Unit exergy cost of main products.

# 5. Conclusions

This preliminary exergy cost analysis allowed the comparison of different routes of vinasse treatment aiming at its energy recovery.

This analysis allowed to visualise and compare production costs, in terms of exergy, of each product of the sugarcane processing plant. The results presented the sugar as the cheapest product, followed by the ethanol, while the clean biogas of Case iii was the most expensive. In addition, this preliminary exergy cost analysis also indicates the impacts in unit exergy costs caused by the introduction of alternative process to treat vinasse with energy recovery.

Furthermore, the results showed that the production of biomethane, as a new product, would be preferable than the production of biogas to be burned in a boiler for electricity production.

# Acknowledgments

The authors would like to thank CAPES, Brazil, CNPq, Brazil [Process: 306303/2014-0; Process: 407175/2018-0 and Process: 429938/2018-7], and UFABC, Brazil.

## References

- [1] FAO Food and Agriculture Organization of the United Nations. FAOSTAT Available at:<<u>http://www.fao.org/faostat/en/#data/QC/visualize</u>> [accessed 13.12.2018].
- [2] Goldember J., Coelho S.T., Nastari P.M., Lucon O., Ethanol learning curve The Brazilian experience. Biomass and Bioenergy 2004;26(3):301-304.
- [3] Coelho J.M., Presentation regarding the EPE and the Renovabio program (in Portuguese). V Symposium of the Post-Graduation Course in Energy of the Federal University of ABC; 2017 Nov 28; Santo André, SP, Brazil.
- [4] Freire W.J., Cortez L.A.B, Sugarcane vinasse (in Portuguese). Guaíba, Brazil: Agropecuaria; 2000.
- [5] Tsatsaronis G., Thermoeconomic analysis and optimization of energy systems. Progress in Energy and Combustion Science 1993;19(3):227-257.
- [6] Pina E.A., Palacios-Bereche R., Chavez-Rodríguez M.F., Ensinas A.V., Modesto M., Nebra S.A., Reduction of process steam demand and water-usage through heat integration in sugar and ethanol production from sugarcane Evaluation of different plant configurations. Energy 2017;138:1263-1280.
- [7] CITROTEC. Vinasse concentration (in Portuguese). 13 SBA Usina em numerous 2012 Available at:<<u>http://www.stab.org.br/13\_sba\_palestras/24\_citrotec\_concentracao\_vinhaca.pdf</u>> [accessed 28.4.2018].
- [8] Palacios-Bereche M.C., Medina-Jimenez A.C., Palacios-Bereche R., Nebra S.A., Comparison between two alternatives for the energy use of vinasse: Concentration-Incineration vs Biodigestion. In: ENCIT 2018: Proceedings of the 17th Brazilian Congress of Thermal Sciences and Engineering; 2018 Nov 25-28; Águas de Lindóia, SP, Brazil.
- [9] Flores-Zavala B.A., Benefiting from the biogas produced in anaerobic biodigesters for producing biomethane and electric energy (in Portuguese) [dissertation]. Santo André, SP, Brazil: Federal University of ABC; 2016.
- [10] Flores-Zavala B., Palacios-Bereche R., Nebra S.A., Exergy and energy analysis of the water scrubbing process applied to biogas upgrading. In: ECOS 2015: Proceedings of the 28th International Conference on Efficiency, Cost, Optimization, Simulation, and Environmental Impact of Energy Systems; 2015 Jun 30 - Jul 3; Pau, France.
- [11] Sosa-Arnao, J. H., 2018. Personal communication. São Paulo.
- [12] Palacios-Bereche M.C., Palacios-Bereche R., Nebra S.A., Comparison through exergy assessment of two alternatives for the energy use of vinasse : Concentration with incineration vs . Biodigestion. In: ECOS 2018: Proceedings of the 31st International Conference on Efficiency, Cost, Optimization, Simulation, and Environmental Impact of Energy Systems; 2018 Jun 17-22; Guimarães, Portugal.
- [13] Dias M.O.D.S., Simulation of the ethanol production process from sugarcane and sugarcane bagasse, aiming at the process integration and production maximisation of energy and bagasse surpluses (in Portuguese) [dissertation]. Campinas, SP, Brazil: University of Campinas; 2008.
- [14] Palacios-Bereche R., Modelling and energy integration of the ethanol production process from sugarcane biomass (in Portuguese). [thesis]. Campinas, SP., Brazil: University of Campinas; 2011.
- [15] Ensinas, A.V., Thermal integration and thermoeconomic optimisation applied to the sugar and ethanol industrial process from sugarcane (in Portuguese). [thesis]. Campinas, SP., Brazil: University of Campinas; 2008.
- [16] Fukushima, N.A., Energy analysis of the integration of a vinasse concentration and incineration system into a sugar and ethanol plant (in Portuguese). [dissertation]. Santo Andre, SP, Brazil: Federal University of ABC; 2016.

- [17] Elia Neto A., Shintaku A., Use and reuse of water and effluent generation. In: Handbook of water reuse and conservation in the sugar-energy agro-industry (in Portuguese). Brasília: National Agency of Water, Environment Ministry, 2009. p. 67–180.
- [18] Elia Neto A., Shintaku A., Good Industrial Practices. In: Handbook of water reuse and conservation in the sugar-energy agro-industry (in Portuguese). National Agency of Water, Brasília: Environment Ministry, 2009, p. 181–256.
- [19] Salomon K.R., Lora E.E.S., Rocha M.H., del Olmo O.A., Cost calculations for biogas from vinasse biodigestion and its energy utilization. Sugar Industry 2011; 4:217-223.
- [20] Leme R.M., Seabra J.E.A., Technical-economic assessment of different biogas upgrading routes from vinasse anaerobic digestion in the Brazilian bioethanol industry. Energy 2017;119:754-766.
- [21] Cozma P., Wukovits W., Mămăligă I., Friedl A., Gavrilescu M., Modeling and simulation of high pressure water scrubbing technology applied for biogas upgrading. Clean Technologies and Environmental Policy 2014;17(2).
- [22] Gallego-Ríos J.M., Effect of processing conditions on the properties of briquettes produced from residues from the ethanol production: Characterisation and analysis of the thermogravimetric behaviour (in Portuguese) [dissertation]. Santo André, SP, Brazil: Federal University of ABC; 2017.
- [23] Cortes-Rodríguez E.F., Nebra S.A., Sosa-Arnao J.H., Experimental efficiency analysis of sugarcane bagasse boilers based on the first law of thermodynamics. Journal of the Brazilian Society of Mechanical Sciences and Engineering 2017;39(3):1033-1044.
- [24] Szargut J., Morris D.R., Steward F.R., Exergy analysis of thermal, chemical and metallurgical processes. New York, USA: Hemisphere Publ. Corp; 1988.
- [25] Ensinas A.V., Nebra S.A., Exergy Analysis as a Tool for Sugar and Ethanol Process Improvement. In: Pélissier G., Calvet A., editors. Handbook of Exergy, Hydrogen Energy and Hydropower Research. Nova Science Publishers. 2009. p. 125-160.
- [26] Modesto M., Nebra S.A., Zemp R.J., A Proposal to Calculate the Exergy of Non Ideal Mixtures Ethanol-Water Using Properties of Excess. In: Proceedings of the 14th European Biomass Conference; 2005 Oct 17-21; Paris, France.
- [27] Lozano M.A., Valero A., Theory of the Exergetic Cost. Energy. Energy 1993;18(9):939-960.
- [28] Silva, M.M., Repowering of power generation systems in the steel industry using thermoeconomic analysis (in Portuguese). [thesis]. Campinas, SP., Brazil: University of Campinas; 2004.

# Appendix A

Figures A.1 and A.2 present the flow sheet diagrams of the conventional production process (A.1) and the vinasse concentration and biogas routes (A.2) depicting the participating process streams.



Fig. A.1. Flow sheet of conventional production process.



Fig. A.2. Flow sheet of concentration and biogas routes.

# Appendix B

Table B1 presents the description of the process streams used in this study.

Deseri	ntion	m	Т	Р	brix,	ethanol,	ex
Descri	phon	(kg/s)	(°C)	(bar)	%	%	(kJ/kg)
1	Sugarcane	138.9	25	1.013	19.14	_	5760
2	Imbibition water	41.67	50	1.013	-	-	54.1
3	Removed impurities	1.727	25	1.013	-	-	-
4	Loss of sucrose	0.3991	32.01	1.013	_	_	_
5	Bagasse	37.77	32.01	1.013	1.66	_	10,055
6	Raw juice	140.7	32.01	1.013	15.41	_	2,742
8	Raw Juice – Sugar production	98.40	32.01	1.013	15.41	_	2,742
9	$SO_2$ for subhitation – sugar production	0.08333	25	1.013	-	_	4 892
10	CaO for liming – sugar production	0.1258	25	1.013	_	_	1.965
11	Water for Ca(OH) <sub>2</sub> preparing –sugar production	1.963	25	1.013	_	_	49.96
12	Vapour from flash – sugar production	1.275	99.02	0.97	_	_	532.4
13	Water for polymer dilution – sugar production	1.458	25	1.013	-	_	49.96
14	Water for filter – sugar production	2.917	25	1.013	_	_	49.96
15	Filter cake – sugar production	3.966	87.55	1.013	-	_	_
16	Water for barometric condenser – sugar production	17.66	30	1.013	-	-	50.13
17	Outlet of barometric condenser – sugar production	18.28	50.38	0.3		-	54.15
18	Clarified juice – concentration	99.63	98.11	1.013	14.76	_	2,658
19	Vegetal Vapour – sugar production	14.06	115.3	1.69	-	_	613.2
20	Condensate of <i>vegetal vapour</i> – sugar production	14.00	22.01	1.09	_ 15 /1	_	97.70
21	Raw Juice – ethanol production	42.2	32.01	1.013	15.41	_	2,742
22	Vapour from flash – ethanol production	0.5363	99.02	0.97	1.00	_	532.4
24	CaO for liming – ethanol production	0.06944	25	1 013	_	_	1 965
25	Water for $Ca(OH)_2$ preparing – ethanol production	1.083	25	1.013	_	_	49.96
26	Water for polymer dilution – ethanol production	0.625	25	1.013	_	_	49.96
27	Water for filter – ethanol production	1.25	25	1.013	_	_	49.96
28	Filter cake – ethanol production	1.681	86.11	1.013	-	_	_
20	Water for barometric condenser - ethanol	6 957	30	1.013			50.13
29	production	0.937	50	1.015	_	—	50.15
30	Outlet of barometric condenser – ethanol	7.201	50.38	0.3	_	_	54.15
21	production Clarified inice and another	42.07	06.45	1.012	14 (7		2 (41
22	Variated juice – must preparation	42.97	90.45	1.013	14.0/	_	2,041
32	$C_{ondensate of vagetal vanour - ethanol production}$	5.852	115.5	1.09	_	_	015.2
34	Exhaust steam – jujce evaporation system	44 18	127.4	2.5	_	_	668.4
	Condensate of exhaust steam – juice evaporation		127.1	2.0			
35	system	42.42	127.4	2.5	-	-	110.7
36	Vegetal vapour for pan 1– crystallisation	11.39	115.3	1.69	-	_	613.2
37	<i>Vegetal vapour for pan 2</i> – crystallisation	1.825	115.3	1.69	-	_	613.2
38	Condensate of vegetal vapour – first effect	7.332	115	1.69	-	-	97.76
39	Condensate of <i>vegetal vapour</i> – second effect	8.035	107.3	1.307	-	-	90.41
40	Condensate of <i>vegetal vapour</i> – third effect	8.737	97.63	0.93	-	_	81.94
41	Condensate of <i>vegetal vapour</i> – fourth effect	9.495	83.27	0.54	-	_	/1.04
42	concentration	298.9	30	1.013	-	_	50.13
	Outlet of barometric condenser – juice						
43	concentration	306.2	50.18	0.16	-	-	54.08
44	Syrup	22.62	55.5	0.16	65	_	11,422
45	Syrup for crystallisation	21.5	55.5	0.16	65	_	11,422
46	Water for centrifuge 1 – crystallisation	1.748	107.4	6	_	_	90.94
47	Water for pan 2 – crystallisation	0.3942	107.4	6	-	_	90.94
48	Water for centrifuge 2 – crystallisation	1.291	107.4	6	-	-	90.94
49	Water for molasses dilution – crystallisation	0.653	107.4	6	-	-	90.94
50	Condensate of <i>vegetal vapour</i> from pan 1	11.39	115	1.69	-	-	97.76
51 52	Undensate of vegetal vapour from pan 2	1.825	115	1.09	-	-	9/./6
52 52	Outlet of harometric condensor – crystallisation	283.3	3U 50.20	0.14	_	_	50.15 54.15
55 54	Wet sugar	293.2 9.408	50.39 60.63	0.10	0	_	54.15 17 506
55	Molasses	6 147	57.68	0.16	73	_	12 824
56	Cold air – sugar drying	4.54	25	1.013	_	_	-
57	Exhaust steam – sugar drying	0.1566	127.4	2.5	_	_	668.4
58	Condensate of exhaust steam – sugar drying	0.1566	127.4	2.5	_	_	110.7

59	Wet air	4.54	25	1.013	-	_	-
60	Dry sugar	9.498	25	1.013	99.9	_	17,537
61	Syrup for must preparation	1.26	55.5	0.16	65	-	11,422
62	Water for must dilution	2.109	25	1.013	-	-	49.96
63	Cooling water for must	521	25	1.013	-	_	49.96
64	Outlet of cooling water	521	30	1.013	-	-	50.13
65	Water for gas separation – fermentation	1.888	25	1.013	-	_	49.96
66	Separated $CO_2$	4.433	30.8	1.013	-	-	-
67	Water for centrifuge – fermentation	13.28	25	1.013	-	_	49.96
68	Y east purge	0.6988	29.78	1.013	-	_	-
69 70	Nutrient for yeast ( $NH_3$ )	0.01557	25	1.013	-	-	19,841
/0 71	$H_2SO_4$ for pH regulation	0.0006944	25	1.013	-	_	1,000
/1 72	Wine	11.02	20.86	1.013	-	- 6 152	49.90
72	Coses concreted distillation	/ 5.45	29.80	1.015	-	0.135	2,165
/3 74	Gases separated – distillation	0.08017	33 25	1.338	-	9.028	26 125
74	Eusel oil 4	0.09807	00 32	1.556	-	25.2	20,135
75	Fusel oil 26	0.008333	90.32 82.28	1.10	-	23.2 83.3	_
70	Pulser on 20	0.02444	02.20	1.10	_	0.210	160.0
79	Vinesse (as diluted solution)	5.07	74.86	1.10	4 20	0.219	109.9
70 70	Villasse (as difuted solution)	02.88	/4.80	1.393	4.39	0.02049	413.7
79 80	Expanse steem distillation	4.007	127 /	2.5	_	95.17	668 /
80 91	Condensate of exhaust steem distillation	15.31	127.4	2.5	_	—	110.7
87	Cooling water distillation	031.5	25	1.013	_	—	10.7
02 92	Outlet of eacling water	931.5	20	1.013	_	—	49.90
03	Vinesse (as solid fuel)	62.00	71.96	1.015	4 20	-	50.15 651 A
04 95	Villasse (as solid luel)	02.88	127 4	1.595	4.39	0.02049	668 4
83	Condensate of exhaust steem	15.44	127.4	2.3	-	_	008.4
86	concentration	13.44	127.4	2.5	-	_	110.7
	Condensate of <i>wagetal</i> waneur 1st offeet of						
87	vinesse concentration system	7.399	122.3	2.139	-	_	105.2
	Condensate of <i>vagatal vanour</i> 2nd effect of						
88	vinesse concentration system	7.825	116.7	1.788	-	_	99.43
	Condensate of <i>wagetal</i> wansur 2rd affect of						
89	vinces concentration system	8.203	110.3	1.449	-	_	93.19
	Condensate of wagetal waneur Ath offect of						
90	vinesse concentration system	8.527	102.8	1.12	-	_	86.35
	Condensate of use at a system 5th offset of						
91	vinesse concentration system	8.788	93.58	0.802	-	_	78.66
	Condensate of userstal usersure 6th offset of						
92	vinesse concentration system	8.967	81.12	0.496	-	_	69.58
	Water for heremetric condensor vinesse						
93	water for barometric condenser – vinasse	252.3	30	1.013	-	_	50.13
	Outlet of herometric condensor vinesse						
94	outlet of balometric condensel – vinasse	261.4	50.38	0.3	-	_	54.15
05	Concentration	4.15	60.07	0.2	65		8 025
95	Electricity for vinesse biodigestion	4.15	00.07	0.2	05	—	0,935
90	Electricity for virias autraction	_	-	_	-	_	9.383
9/	Electricity for juice extraction	_	-	-	_	_	9200 450*
90	Electricity for juice treatment – sugar production	_	-	_	-	_	450
99 100	Electricity for juice treatment – ethanol production	_	-	-	_	_	430
100	Electricity for sugar crystallisation	_	-	_	-	_	900 1800*
101	Electricity for sugar draing	—	_	_	_	—	150
102	Electricity for formentation	_	-	_	-	_	130 600*
105	Electricity for distillation	_	-	-	_	_	450*
104	Decence for calf consumption	1 990	22.01	-	-	_	430
105	Dagasse for bailer	1.009	32.01 22.01	1.015	1.00	_	10,035
100	Dagasse for bolier	55.19	32.01 22.01	1.013	1.00	_	10,055
107	Steam an antal in hailan	0	52.01	1.015	1.00	_	10,035
108	Steam generated in boller	A	520 127.4	05	_	-	1,402
109	Condensation of managements (notices to be silve)	В	127.4	2.5	-	-	008.4
110	Condensates of process steam (return to boller)		102	2.09	_	-	83./3
111	Flastricity for reverse 1	D	25	2.09	-	-	50.07 E*
112	Electricity for pump 1 – cogeneration system	-	-	-	-	—	Е г*
113	Electricity for pump 2 – cogeneration system	-	-	-	-	_	F
114	Biourgested vinasse	62.57	30	1.013	-	_	130.3
115	Biogas	0.5138	30	1.013	-	-	18,499
116	Electricity for desulphurisation	-	- 25	-	-	_	56.62
117	Effluent from desulphurisation	0.0767	35	1.1	-	-	-
118	Clean blogas	0.4371	35	1.1	-	-	19,795
119	Air – biogas purification	1.46	25	1.01	-	-	-

120	Electricity for purification	_	_	_	_	-	344.7*
121	Make-up water – biogas purification	2.404	20	1.1	-	_	50.14
122	Water purge	2.404	15	1.1	-	_	50.68
123	Exhaust stream – biogas purification	1.723	15	1.1	_	_	64.59
124	Biomethane	0.1748	15	10	_	_	49,116

\* Total exergy; A: 76.69 kg/s for Case i, 85.91 kg/s for Case ii, 78.93 kg/s for Case iii, 76.69 kg/s for Case iv; B: 59.65 kg/s for Case iii, 57.26 kg/s for Case iii, 71.17 kg/s for Case iii, 57.26 kg/s for Case iii, 57